

## Photoionization and Recombination

At equilibrium:

$$\begin{aligned} \# \text{ of ionizations per unit volume} \\ = \# \text{ of recombinations per unit volume} \end{aligned}$$

$$N_{H^0} \int_{\nu_0}^{\infty} \frac{4\pi J_{\nu}}{h\nu} a_{\nu}(H^0) d\nu = N_e N_p \alpha(H^0, T)$$

$J_{\nu}$  : mean intensity of radiation,  $\text{erg s}^{-1} \text{cm}^{-2} \text{Hz}^{-1} \text{steradian}^{-1}$

$N_{H^0}$  = No. of neutral H atom,  $\text{cm}^{-3}$

$a_{\nu}$  = ionization cross section,  $\text{cm}^2$

$\alpha$  = recombination coefficient,  $\text{cm}^3 \text{s}^{-1}$

$$a_{\nu}(Z) = \frac{A_0}{Z^2} \left( \frac{\nu_1}{\nu} \right)^4 \frac{e^{4 - [(4 \tan^{-1} \epsilon)/\epsilon]}}{1 - e^{-2\pi/\epsilon}} \quad \text{for } \nu > \nu_1$$

$$\text{where } A_0 = \frac{2^8 \pi}{3 e^4} \left( \frac{1}{137.0} \right) \pi a_0^2 = 6.30 \times 10^{-18} \text{cm}^2$$

$$\epsilon = \sqrt{\frac{\nu}{\nu_1} - 1}$$

$$\text{and } h\nu_1 = Z^2 h\nu_0 = 13.6 Z^2 \text{eV}$$

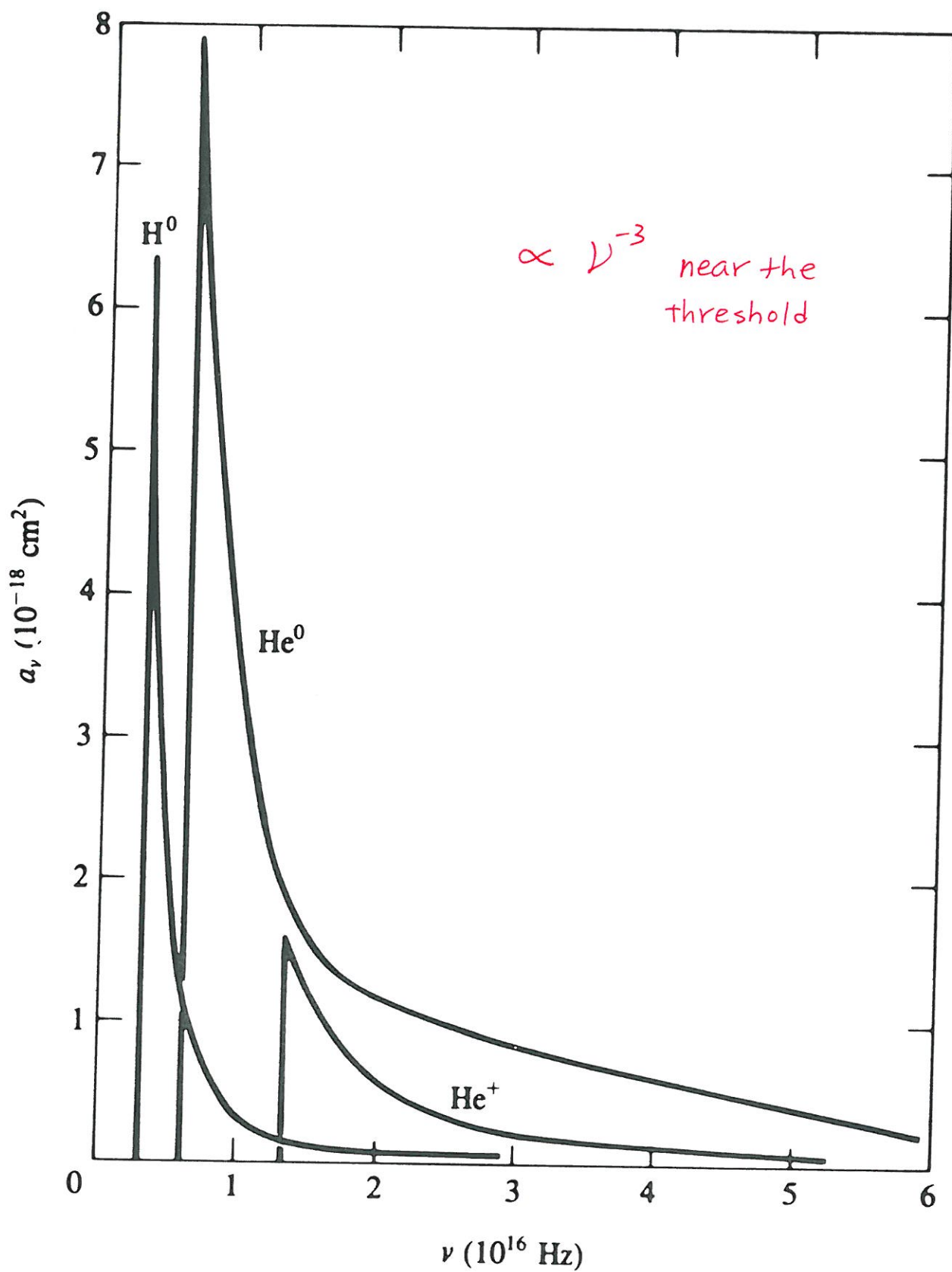


FIGURE 2.2  
Photoionization absorption cross sections of  $\text{H}^0$ ,  $\text{He}^0$ , and  $\text{He}^+$ .

Recombination coefficient to a specific level  $n^2L$

$$\alpha_{n^2L}(H^0, T) = \int_0^\infty v \sigma_{nL}(H^0, v) f(v) dv,$$

$$\text{where } f(v) = \frac{4}{\sqrt{\pi}} \left( \frac{m}{2kT} \right)^{3/2} v^2 e^{-mv^2/2kT}$$

is the Maxwell-Boltzmann distribution function for the electrons.

$\sigma_{nL}(H^0, v)$  is the recombination cross section to the term  $n^2L$  for electrons with velocity  $v$ .

$$\sigma \propto v^{-2} \Rightarrow \alpha \propto T^{-1/2}$$

The total recombination coefficient

$$\alpha_A = \sum_{n,L} \alpha_{n^2L}(H^0, T) \quad (\text{Table 2.1})$$

A typical recombination time is  $\frac{1}{N_e \alpha_A}$

$$\alpha_A = 4.18 \times 10^{-13} \text{ cm}^3 \text{ sec}^{-1} \text{ at } 10^4 \text{ K}$$

$$\therefore \text{recombination time is } \frac{7.6 \times 10^4}{N_e} \text{ yr}$$

Approximately, recombination time  $\approx \frac{10^5}{N_e} \text{ yr.}$

What happens after a recombination?

How long does it take to photoionize an  $H^0$ ?



Recombination  $\rightarrow$   $H^0$  at the excited level  $nL$

$A_{nL, n'L'}$  : transition probability from  $nL$  to  $n'L'$   
 $\sim 10^4$  to  $10^8 \text{ sec}^{-1}$

The mean lifetime of the excited level  $nL$  is

$$\tau_{nL} = \frac{1}{\sum_{n' < n} \sum_{L' = L \pm 1} A_{nL, n'L'}}$$

$\approx$  of order  $10^{-4}$  to  $10^{-8} \text{ sec}$ .

$H^0$  at  $nL$  makes permitted one-photon downward transitions to the ground state  $1^2S$ .

The  $2^2S$  level is the only exception, for which two-photon emission is needed, and  $A_{2^2S, 1^2S} = 8.23 \text{ sec}^{-1}$ .  
The life time of  $2^2S$  is  $0.12 \text{ sec}$ .

All cascade down to  $1^2S$  in  $\ll 1 \text{ sec}$ .

At  $5 \text{ pc}$  from an  $O6$  star with an ionizing luminosity of  $10^{48.7} \text{ photons/sec}$ ,  
the photoionization rate is  $\alpha(H^0) \frac{10^{48.7}}{4\pi d^2} \text{ sec}^{-1} \approx 10^{-8} \text{ sec}^{-1}$   
 $\alpha(H^0) \approx 6 \times 10^{-18} \text{ cm}^2$   
 $d = 5 \text{ pc} = 1.5 \times 10^{19} \text{ cm}$

The lifetime of  $H^0$  against photoionization is  
 $10^8 \text{ sec} \sim 3 \text{ yr}$